## ™ NATIONAL TALENT SEARCH EXAMINATION (NTSE-2017) STAGE -1

STATE: PUNJAB PAPER: SAT

Date: 05/11/2017

Max. Marks: 100 SOLUTIONS Time allowed: 90 mins

**51.** A body of mass 1 kg initial at rest is moved by a horizontal force of  $0.5 \, \mathrm{N}$  on a smooth frictionless table. The work done by the force in  $10 \, \mathrm{sec}$ . is

 $(1)\ 10.5 J$ 

(2) 12.5 J

(3) 20 J

(4) 22 J

Ans. (2)

**Sol.**  $W = F \times s$ 

F = ma  $a = \frac{F}{m} = \frac{0.5}{1} \text{ m/s}^2$ 

 $s = ut + \frac{1}{2}at^2$ 

 $s = 0 + \frac{1}{2} \times 0.5 \times 10 \times 10 = 25 \text{ m}$ 

 $W = 0.5 \times 25 = 12.5 J$ 

**52.** A boy hears an echo of his own voice from a distant hill after 2 seconds. The speed of sound in air is 350 m/s. The distance of the hill from the boy is

(1) 175 m

(2) 200 m

(3) 350 m

(4) 250 m

Ans. (3)

**Sol.**  $v = \frac{2s}{4}$ 

 $350 = \frac{2s}{2} = 350 \text{ m}$ 

**53.** An electric heater consists of 20 m length of manganin wire of  $0.23 \text{ m}^2$  cross sectional area. Wattage of heater when potential difference across the heater is 200 V is

 $(1) 10^6 W$ 

(2) 100 W

 $(3) 10^9 W$ 

 $(4) 10^3 W$ 

Ans. (3)

**Sol.**  $P = \frac{V^2}{R}$ 

 $R = \frac{\rho \ell}{A} = \frac{48.2 \times 10^{-8} \times 20}{0.23}$ 

 $P = \frac{V^2}{R} = \frac{200 \times 200 \times 0.23 \times 10^8}{48.8 \times 20} \approx 9.5 \times 10^8 \approx 10^9 \text{ Watts}$ 

**54.** A beam of white light when passes through a glass prism, a spectrum is observed. But when same beam of light passes through hollow grass prism then

(1) Spectrum is same

(2) Spectrum become brighter

(3) There will be no spectrum

(4) Colours of spectrum reversed

Ans. (3)

**Sol.** There will be no spectrum.

- **55.** When a current carrying conductor is placed in a direction parallel to the magnetic field, force on conductors is
  - (1) Zero

- (2) 100 N
- (3) 10 N

(4) 1000 N

Ans. (1)

- **Sol.** When a current carrying conductor is placed in a direction parallel to the magnetic field, force on conductors is
- **56.** An object of size 4 cm placed perpendicular to the principal axis of concave mirror. The distance of the object from the mirror equals radius of curvature. The size of the image will be
  - (1) 1 m

- (2) 2 m
- (3) 3.5 m
- (4) 4 cm

Ans. (4)

**Sol.** u = R

 $h_1 = 4 \text{ cm}$ 

 $h_2 = ?$ 

 $\frac{1}{v} + \frac{1}{(-u)} = \frac{1}{(-f)}$ 

 $\frac{1}{v} = \frac{1}{u} - \frac{1}{f}$ 

 $\frac{1}{v} = \frac{1}{u} - \frac{1}{u/2} \Rightarrow \frac{1}{v} = \frac{1-2}{u} \Rightarrow \frac{1}{v} = \frac{-1}{u}$ 

v = -u

 $h_2 = \frac{-v}{u} \times h = 1 \times 4 = 4 \text{ cm}$ 

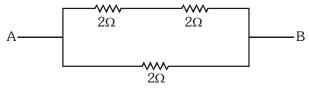
As the object is placed on centre of curvature, the size of image will be equal to the size of object.

- **57.** The most important safety method used for protecting home appliances from short circuiting or overloading is
  - (1) Earthing
- (2) Use of fuse
- (3) Use of stabilizer
- (4) Use of electric meter

Ans. (2)

Sol. Use of fuse

**58.** The equivalent resistance between the points A and B in the circuit as shown in the figure given below is



(1) 1 ohm

(2) less than 1 ohm

(3) more than 1 ohm

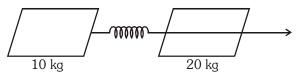
(4) 8 ohm

Ans. (3)

Sol.  $R_{eq} = A^{\bullet}$ 

 $R_{eq} = \frac{R_1 R_2}{R_1 + R_2} = \frac{8}{6} = \frac{3}{2} \Omega$ 

The masses of 10 kg and 20 kg respectively are connected by massless spring. A force of 200 N acts on the 20 **59**. kg mass. At the instant shown, the 10 kg mass has an acceleration of 12 m/sec<sup>2</sup>. What is the acceleration of 20 kg mass?



- (1)  $12 \text{ m/sec}^2$
- (2) 4 m/sec<sup>2</sup>
- (3) 20 m/sec<sup>2</sup>
- $(4) 5 \text{ m/sec}^2$

Ans. (2)

**Sol.** 
$$F = m_1 a_1 + m_2 a_2$$

$$200 = 10 \times 12 + 20a_9$$

$$200 - 120 = 20a_2$$

$$80 = 20a_2$$

$$a_2 = 4 \text{ m/s}^2$$

- **60.** A bullet of mass 0.1 kg is fired with a speed of 100 m/sec., the mass of gun is 50 kg. The velocity of recoil is
  - (1) 0.2 m/sec
- $(2) 0.1 \, \text{m/sec}$
- $(3) 0.5 \, \text{m/sec}$
- $(4) 0.05 \, \text{m/sec}$

Ans. (1)

**Sol.** 
$$m_1u_1 + m_2u_2 = m_1v_1 + m_2v_2$$

$$0 = 0.1 \times 100 - 50 v_2$$

$$v_2 = \frac{10}{50} = 0.2 \text{ m/s}$$

- 61. Which of the following frequency of sound can be generated by a vibrating simple pendulum as well as by the vibrating vocal cords of a rhinocerous?
  - (1) 8 Hz

- (2) 25 Hz
- (3) 10 Hz
- (4) 15000 Hz

Ans. (3)

**Sol.** 10 Hz

- **62.** Which of the following is not an example of biomass energy source?
  - (1) Wood
- (2) Garbage
- (3) Atomic energy
- (4) Coal

Ans. (3)

**Sol.** Atomic energy

- A needle placed 45 cm from a lens forms an image on a screen placed 90 cm on the other side of lens, focal length and type of lens is
  - (1) +10cm, Convex lens
- (2) +30 cm, Convex lens (3) +30 cm, Concave lens (4) +10 cm, Concave lens

Ans. (2)

**Sol.** 
$$u = 45 \text{ cm}$$

$$v = 90 \text{ cm}$$

$$\frac{1}{v} - \frac{1}{u} = \frac{1}{f}$$
  $\frac{1}{90} - \frac{1}{(-45)} = \frac{1}{f}$ 

$$\frac{1}{90} + \frac{1}{45} = \frac{1}{f} \qquad \frac{1+2}{90} = \frac{1}{f}$$

$$f = \frac{90}{3} = 30 \text{ cm}$$

As the image is formed on the other side of lens, it is a convex lens.

<b>64</b> .	If $1.5$ moles of oxygen gas combines with solid Aluminium to form $\mathrm{Al_2O_3}$ . What is the mass of Aluminium used in the master $2$					
	in the reaction?					
	(1) 27 gm	(2) 54 gm	(3) 40.5 gm	(4) 81 gm		
Ans.	(2)					
Sol.						
	3 moles of $O_2$ combines with $Al = 4$ moles					
	1 mole of $O_2$ combines with $Al = \frac{4}{3}$ moles					
	1.5 moles of $O_2$ combines with $Al = \frac{4}{3} \times 1.5 = 2$ moles					
	Mass of Al in $Al_2O_3 = 2 \times 2$	7 = 54  g				
<b>65</b> .				ged to green. He checked the		
	pH again after dissolving a p	pinch of common salt in wa	ater. The colour of pH pape			
	(1) Green	(2) Yellow	(3) Red	(4) Blue		
Ans.	(1)					
Sol.	pH paper shows green color			re.		
66.	On electrolysis of brine solut					
	(1) Na and Cl <sub>2</sub>	(2) H2, O2 and Cl2	(3) $H_2$ , $Cl_2$ and NaOH	(4) NaOH, $Cl_2$ and $O_2$		
Ans.	(3)					
Sol.	$\underbrace{NaCl + H_2O}_{\text{Brine}} \xrightarrow{\text{electrolysis}} N$	$IaOH + H_2 \uparrow + Cl_2 \uparrow$				
<b>67</b> .	What is the valency of an ele	ement having atomic numb	er 18?			
	(1) 2	(2) 8	(3) Zero	(4) 6		
Ans.	(3)					
Sol.	Atomic no. 18 is a noble gas	s i.e. Argon has valency zer	·o.			
<b>68</b> .	Number of molecules presen	nt in 18g and 18u of water	respectively.			
	(1) $6.023 \times 10^{23}$ and 1		(2) $6.023 \times 10^{23}$ and 18			
	(3) 18 and $6.023 \times 10^{23}$		(4) 1 and $6.023 \times 10^{23}$			
Ans.	(1)					
Sol.	18 g of water contians 1 mo	ole of molecules of water =	$6.023 \times 10^{23}$ molecules.			
	18 u of water contians 1 mo	olecule of water because m	olecular mass of $H_2O = 18$	3 u.		
<b>69</b> .	By adding acetic acid to soli milky. What are X and Y.	d "X" a colourless and odo	urless gas "Y" is evolved. 1	The gas "Y" turns lime water		
	(1) X is Sodium carbonate ar	nd Y is $CO_2$ .	(2) X is Sodium hydroxide	e and Y is CO <sub>2</sub> .		
	(3) X is Sodium acetate and	Y is CO <sub>2</sub> .	(4) X is Sodium bicarbona	ate and Y is $SO_2$ .		
Ans.	(1)					
Sol.	A antica and A Calid V					
	Acetic acid + Solid $X \longrightarrow C$	CO <sub>2</sub> (gas Y)				
	Acetic acid + Solid $X \longrightarrow C$ Milky $\longleftarrow$ Lime $V$					
	•	water				
	Milky ←— Lime v	water onates or $\longrightarrow$ $CO_2$				

 $\mathsf{CaCO}_{\scriptscriptstyle{3}} \longleftarrow \mathsf{Lime} \ \mathsf{water}$ 

Milky

 $Ca(OH)_2$ 

- **70.** High melting point of a compound indicates
  - (1) Strong intermolecular forces.

(2) Kinetic energy of molecules is more.

(3) Speed of molecules is more.

(4) Compound can diffuse easily.

Ans. (1)

- **Sol.** Strong intermolecular forces cause high melting point of a compound. e.g. Ionic compounds are strong with high melting point.
- **71.** A metal on heating in presence of air gives compound which is soluble in water and have high melting point. The metal is:
  - (1) Calcium
- (2) Carbon
- (3) Silicon
- (4) Iron

Ans. (1)

**Sol.** Group 2 element generally have high melting point.

$$Ca + O_2 \longrightarrow CaO$$

 $CaO + H_2O \longrightarrow Ca(OH)_2$  soluble in water.

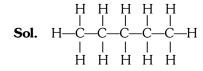
- 72. An element which is essential constituent of all organic compounds belongs to which group in Periodic Table.
  - (1) Group 2
- (2) Group 14
- (3) Group 16
- (4) Group 17

Ans. (2)

- **Sol.** Carbon is essential constituent of all organic compounds belongs to group 14.
- **73.** How many covalent bonds are present in Pentane?
  - (1) 8

- (2) 10
- (3) 16
- (4) 14

Ans. (3)



16 Covalent bonds

- **74.** Which one of the following is not an isoelectronic with Neon atom?
  - $(1)_{8}O^{-2}$

- (2) <sub>11</sub>Na<sup>+1</sup>
- $(3)_{o}F^{-1}$
- (4)  $_{12}{\rm Mg}^{+1}$

Ans. (4)

**Sol.** Neon atomic no. = 10 have 10 electrons.

 $_{12}{
m Mg^{+1}}$   $\Rightarrow$  has 11 electrons and is not isoelectronic (same no. of electrons) with Neon.

- **75.** The formula of sulphate of element X is  $X_2(SO_4)_3$ . The formula of nitride of element X is
  - (1)  $X_{2}N$

- (2)  $XN_{2}$
- (3) XN
- $(4) X_2 N_3$

Ans. (3)

**Sol.** Valency of X in  $X_2(SO_4)_3 = 3$ .

Nitride =  $N^{3-}$ 

Formula of nitride of X



**76.** Three test tubes containing:

Test Tube X: 5ml of HCl

Test Tube Y; 5ml of HNO<sub>3</sub>

Test Tube Z: 15ml of HCl + 5ml of HNO<sub>3</sub>

A small piece of metal was added in all the three test tubes. It dissolves only in Test Tube Z. The metal is:

(1) Al

(2) Au

(3) Cu

(4) Ag

Ans. (2)

**Sol.** In test tube Z: 15 ml of HCl + 5 ml of  $HNO_3$ .

Ratio 3:1 of HCl and HNO $_3$  is known as aqua regia, which is used to dissolve noble metals like Au (Gold) and Pt (Platinum).

77. Match the column I with the column II.

	Column I		Column II
(i)	Renal artery	(a)	It stores the urine untill it is released through the Urethra.
(ii)	Kidney	(b)	It passes urine from kidney to urinary bladder.
(iii)	Ureter	(c)	It filters the blood and forms urine.
(iv)	Urinary bladder	(d)	It bring blood to the kidney for filtering waste.

(1) i-d, ii-c, iii-b, iv-a

(2) i-a, ii-b, iii-c, iv-d

(3) i-b, ii-a, iii-d, iv-c

(4) i-c, ii-b, iii-a, iv-b

Ans. (1)

**Sol.** (i) Renal artery – It bring blood to the kidney for filtering waste.

- (ii) Kidney It filters the blood and forms urine.
- (iii) Ureter It passes urine from kidney to urinary bladder.
- (iv) Urinary bladder It stores the urine untill it is released through the Urethra.
- **78.** Winch one of the following is the function of the enzymes of Pancreatic Juice?
  - (1) trypsin digests protein and lipase digests carbohydrates
  - (2) trypsin digests emulsified fats and lipase digests proteins
  - (3) trypsin and lipase digest fats
  - (4) trypsin digests proteins and lipase digests emulsified fats.

Ans. (4)

**Sol.** Pancreatic juice contains protein digesting enzyme trypsin. It also has lipase which acts on emulsified fats.

**79.** Which of the following is not a correct pair?

(1) Adrenaline: Piluitary gland

(2) Testosterone: Testes

(3) Estrogen: Ovary

(4) Thyroxine: Thyroid gland

Ans. (1)

**Sol.** Adrenaline is released by adrenal gland.

80. Pine and Deodar are the example of

(1) Gymnosperms

(2) Pteridophyta

(3) Thallophyta

(4) Bryophyta

Ans. (1)

**Sol.** Pine and deodar are the examples of gymnosperms.

Khadins, Bundhis and Ahars are ancient structures that are example of (1) Grain storage (2) Wood storage (3) Water harvesting (4) Soil conservation Ans. (3) **Sol.** Khadins, Bundhis and Ahars are the traditional methods of water harvesting. **82.** Read the following statements and select the correct option Statement I: Snails and Mussels are Molluscans. Statement II: Sea Urchins and Scorpions are Echinodermates (1) only statement I is true (2) only statement II is true (3) Both statements I & II are true (4) Both statements I & II are false Ans. (1) **Sol.** Snails and mussels are molluscans. Sea urchins are echinoderms and scorpions are arthropods. the life process of the cell. (1) Golgi Apparatus (2) Nucleus (3) Lysosome (4) Ribosome Ans. (2) **Sol.** Nucleus is a double membrane structure that directs the life process of the cell. Nucleus is known as the director of cell. **84.** A feature that is common in yeast, amoeba, paramecium is (1) They all are multicellular (2) They all reproduce by budding (3) They all reproduce by binary fission (4) They all are unicellular Ans. (4) **Sol.** Yeast is unicellular organism and reproduce by budding. Amoeba and paramecium are unicellular organisms and reproduce by binary fission.

Ans. (2)

(1) Red blood cells

**85**.

**Sol.** Plasma forms 55% of the blood and formed elements (RBC, WBC & Platelets) form 45% of the blood.

**86.** Homologous organs are.....

(1) Dissimilar in origin, similar in function.

(2) Dissimilar in origin and dissimilar in function.

(4) White blood cells

 $(3) \, Similar \, in \, origin \, and \, similar \, in \, function. \\$ 

(4) Similar in origin and dissimilar in function.

(3) Blood protein

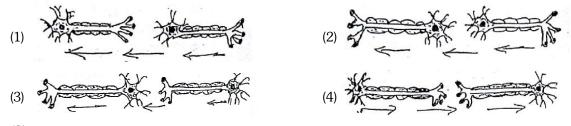
Ans. (4)

**Sol.** Homologous organs are those organs that have basic similar structure but have become modified to perform different functions.

**87.** What is the correct direction of flow of electrical impulses in nerve cells?

Which blood constituent makes up more of the volume of blood?

(2) Plasma



Ans. (3)

**Sol.** Electrical impulse is received by dendrites and is conducted to Axon. Synapse is the junction between dendrite of one neuron and axon of other enuron.

- **88.** You observed a slide of animal tissue and observed (i) long cylindrical and unbranched cells (ii) They had dark and light bands. The tissue could be of
  - (1) Unstriated muscle fibres

(2) Neurons

(3) Striated muscle fibres

(4) Cardiac muscle fibres

Ans. (3)

- **Sol.** Striated or skeletal muscle fibres are long, cylindrical, unbranched with alternate light and dark bands (striations)
- **89.** Which organ is infected when a person suffers from Jaundice?
  - (1) Bones
- (2) Liver
- (3) Lungs
- (4) Nervous system

Ans. (2)

- **Sol.** Liver is affected in Jaundice.
- **90.** The animals having jointed legs belongs to phylum
  - (1) Annelida
- (2) Arthropoda
- (3) Mollusca
- (4) Nematoda

Ans. (2)

- **Sol.** Arthropoda phylum includes animals having jointed legs and segmented bodies.
- **91.** If p + q + r = 0 then the value of  $\frac{2p^2(q+r) + 2q^2(p+r) + 2r^2(p+q)}{pqr}$  will be
  - (1) 3pqr
- (2)  $\frac{1}{pqr}$
- (3) 6

(4) –6

Ans. (4)

**Sol.** 
$$\frac{2p^2(q+r) + 2q^2(p+r) + 2r^2(p+q)}{pqr}$$

$$= \frac{2p^2(-p) + 2q^2(-q) + 2r^2(-r)}{pqr} \ [\because \ p+q+r=0]$$

$$= \frac{-2[p^3 + q^3 + r^3]}{pqr} = \frac{-2(3pqr)}{pqr} \text{ [If } p + q + r = 0, p^3 + q^3 + r^3 = 3pqr]}$$

- = -6
- **92.** A circle is inscribed in a square of side 2.5cm. Another circle is circumscribing this square. The ratio of areas of outer circle and inner circle is
  - (1)  $1:\sqrt{2}$
- (2)  $\sqrt{2} : 1$
- (3) 2 : 1
- $(4) \sqrt{3} : 1$

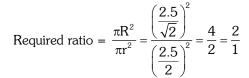
Ans. (3)

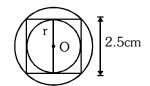
**Sol.** Let radius of inner circle of r and that of outer circle be R.

$$2r = 2.5$$

$$r = \frac{2.5}{2}$$

And 
$$2R = 2.5\sqrt{2} \Rightarrow R = \frac{2.5}{\sqrt{2}}$$





**93.** If  $x = \frac{1}{\sqrt{3}-1}$  then find the value of  $4x^3 + 2x^2 - 8x - 3$ .

(1) 0

(2)2

(3) -2

(4)  $\sqrt{2}$ 

Ans. (1)

**Sol.**  $x = \frac{1}{(\sqrt{3}-1)} = \frac{\sqrt{3}+1}{2}$ 

Then  $x^3 = \left(\frac{\sqrt{3}+1}{2}\right)^3 = \frac{3\sqrt{3}+1+9+3\sqrt{3}}{8} = \frac{10+6\sqrt{3}}{8}$ 

 $x^{2} = \left(\frac{\sqrt{3} + 1}{2}\right)^{2} = \frac{4 + 2\sqrt{3}}{4} = \frac{2(2 + \sqrt{3})}{4} = \frac{2 + \sqrt{3}}{2}$ 

 $\text{Then } 4x^3 + 2x^2 - 8x - 3 = \frac{4 \times 2 \left(5 + 3\sqrt{3}\right)}{8} + \frac{2 \left(2 + \sqrt{3}\right)}{2} - 8 \left(\frac{\sqrt{3} + 1}{2}\right) - 3$ 

 $= (5+3\sqrt{3})+(2+\sqrt{3})-4(\sqrt{3}+1)-3$ 

 $= 5 + 3\sqrt{3} + 2 + \sqrt{3} - 4\sqrt{3} - 4 - 3$ 

= (

**94.** How many revolutions will a circular wheel of radius r unit will make to cover a distance of 100 times its diameter?

(1)  $\frac{100}{\pi}$ 

(2)  $100\pi$ 

(3)  $\frac{\pi}{100}$ 

(4)  $\frac{50}{\pi}$ 

Ans. (1)

**Sol.** Distance to be covered = 100 (2r) = 200r

Distance covered in 1 revolution =  $2\pi r$ 

No. of revolutions =  $\frac{200r}{2\pi r} = \frac{100}{\pi}$ 

**95.** If  $tan\phi + cot\phi = 1$  then the value of  $sin\phi + cos\phi$  will be equal to (where  $\phi$  is an acute angle)

(1) 0

(2)  $\sqrt{2}$ 

(3)  $\sqrt{3}$ 

 $(4)\ 1$ 

Ans. (3)

**Sol.**  $\tan \phi + \cot \phi = 1$  (given)

 $\frac{\sin\phi}{\cos\phi} + \frac{\cos\phi}{\sin\phi} = 1$ 

 $\Rightarrow \frac{\sin^2 \phi + \cos^2 \phi}{\sin \phi \cos \phi} = 1$ 

 $\Rightarrow \sin \phi \cos \phi = 1$ 

..... (i)

Let  $\sin \phi + \cos \phi = x$ 

Then  $[\sin \phi + \cos \phi]^2 = x^2$ 

 $\Rightarrow \sin^2 \phi + \cos^2 \phi + 2\sin \phi \cos \phi = x^2$ 

 $\Rightarrow$  1 + 2 =  $x^2$ 

[By equation (i)]

 $\Rightarrow x^2 = 3$ 

 $\Rightarrow x = \sqrt{3}$ 

**96.** If pth term of an AP is q and qth term is p then mth term of this AP will be :

(1) 
$$p + q + m$$

(2) 
$$p + q - m$$

$$(3) p - q - m$$

$$(4) p - q + m$$

Ans. (2)

**Sol.** 
$$T_p = q \Rightarrow a + (p-1)d = q$$

$$T_q = p \Rightarrow a + (q - 1)d = p$$

On subtracting both equations

$$\Rightarrow [p-1-q+1]d = q-p$$

$$(p-q)d = -(p-q)$$

$$d = -1$$

Then 
$$a + (p - 1)(-1) = q$$
  $\Rightarrow d = q + p - 1$ 

$$\Rightarrow d = q + p - 1$$

$$T_{m} = a + (m - 1) d$$

$$= q + p - 1 + (m - 1) (-1)$$

$$= q + p - 1 - m + 1$$

$$T_m = q + p - m$$

Two dice are thrown. Find the probability that sum of numbers of both up sides of both dice is a perfect square. **97**.

(1) 
$$\frac{1}{6}$$

(2) 
$$\frac{7}{36}$$

(3) 
$$\frac{5}{36}$$

Ans. (2)

**Sol.** Sum of numbers on both dice should be 1, 4, 9, 16, 25 .....

Favourable outcomes =  $\{(1, 3), (2, 2), (3, 1), (3, 6), (4, 5), (5, 4), (6, 3)\}$ 

Total outcomes = 36

Required probability =  $\frac{7}{36}$ 

**98.** From the top of tower of h m high, the angles of depression of two objects, which are in line with the foot of the tower are  $\alpha$  and  $\beta$  ( $\beta > \alpha$ ). Find the difference between two objectives.

(1) 
$$h(\tan \alpha - \tan \beta)$$

(2) 
$$h(\cot \alpha - \tan \beta)$$

(3) 
$$h(\cot \alpha - \cot \beta)$$

(4) 
$$h(\cot \alpha + \cot \beta)$$

Ans. (3)

**Sol.** As shown in figure

Let 
$$BC = x$$
 units

$$BD = y \text{ units}$$

In  $\triangle ABC$ 

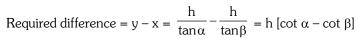
$$\tan \beta = \frac{h}{x}$$

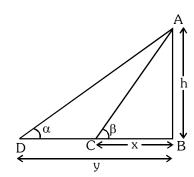
$$\Rightarrow x = \frac{h}{\tan \beta}$$

In AADB

$$\tan \alpha = \frac{h}{y}$$

$$y = \frac{h}{\tan \alpha}$$





- **99.** If the distance between the points (4, q) and (1, 0) is 5 units then the value of q is -
  - (1) 4

(4) 0

Ans. (3)

**Sol.** Distance = 5

$$\Rightarrow \sqrt{(4-1)^2 + (q-0)^2} = 5$$

$$\Rightarrow$$
 3<sup>2</sup> + q<sup>2</sup> = 5<sup>2</sup>

$$\Rightarrow$$
 q<sup>2</sup> = 16

$$\Rightarrow$$
 q =  $\pm 4$ 

**100.** The area of an equilateral triangle is  $49\sqrt{3}$  cm<sup>2</sup>. Taking each vertex as centre, circles are described with radius equal to half the length of the side of the triangle. Find the area of the triangle. Find the area of the part of the

triangle which is not included in these circles. ( $\sqrt{3} = 1.73$ ,  $\pi = \frac{22}{7}$ )

- (1) 84cm<sup>2</sup>
- (2) 77.7cm<sup>2</sup>
- (3) 7.77cm<sup>2</sup>
- (4) 70.7cm<sup>2</sup>

Ans. (3)

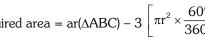
**Sol.** Let AB = 2r

$$ar(\Delta ABC) = 49\sqrt{3}$$

$$\frac{\sqrt{3}}{4}(2r)^2 = 49\sqrt{3}$$

$$\Rightarrow \sqrt{3} r^2 = 49\sqrt{3}$$
$$r = 7 cm$$

Required area =  $ar(\Delta ABC) - 3\left[\pi r^2 \times \frac{60^\circ}{360^\circ}\right]$ 

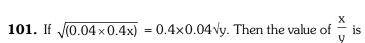




$$= 49\sqrt{3} - \frac{22}{7} \times \frac{1}{2} \times 49$$

$$= 49\sqrt{3} - 77$$

$$= 7.77 \text{ cm}^2$$



- (1) 0.0016
- (2) 0.16
- (3) 0.016
- (4) 1.6

Ans. (3)

**Sol.**  $\sqrt{(0.04)(0.4x)} = (0.4)(0.04)\sqrt{y}$ 

squaring both side then

$$(0.04)(0.4x) = (0.4)^2(0.04)^2y$$

$$\Rightarrow$$
 x = (0.4) (0.04)y

$$\Rightarrow \frac{x}{y} = (0.4)(0.04) = 0.016$$

**102.** One litre of water weighs 1 kg. How many cubic millimetres of water weigh 0.1 gm?

(1) 100

- (2) 10
- (3) 1

(4) 0.1

Ans. (1)

**Sol.** Weight of 1000g water = 1 litre =  $10^6 \text{ mm}^3$ 

So, weight of 1g water =  $\frac{10^6}{10^3}$ mm<sup>3</sup> =  $10^3$ mm<sup>3</sup>

So, weight 0.1g water =  $(0.1)(10^3)$  mm<sup>3</sup>

 $= 100 \text{ mm}^3$ 

**103.** If  $x = \sqrt[3]{2\frac{93}{125}}$ , then the value of x is :

- (1)  $2\frac{1}{5}$
- (2)  $1\frac{2}{5}$
- (3)  $3\frac{4}{5}$

(4)  $4\frac{1}{5}$ 

Ans. (2)

**Sol.**  $x = \sqrt[3]{2\frac{93}{125}}$ 

 $\Rightarrow x = \sqrt[3]{\frac{343}{125}} = \frac{7}{5}$ 

 $\Rightarrow x = 1\frac{2}{5}$ 

104. If x men can do a piece of work in 8 days and (x + 4) men can do the same work in 6 days then x is equal to :

(1) 10

(2)6

(3) 12

(4) 24

Ans. (3)

**Sol.** x men can do work in 8 days

then 1 man can do it in 8X days

(x + 4) men can do work in 6 days

then 1 man can do in 6(x + 4) days

So 8x = 6(x + 4)

$$2x = 24$$

$$\Rightarrow x = 12$$

**105.** If  $x^2 + y^2 + z^2 = r^2$  where  $x = r \sin A \cos B$ ,  $y = r \sin A \sin B$  then Z has one of the following values :

- (1) r sin B
- (2) r cos A
- (3) r tan A cos B
- (4) r tan A tan B

Ans. (2)

**Sol.**  $x^2 + y^2 = r^2 \sin^2 A \cos^2 B + r^2 \sin^2 A \sin^2 B$ 

$$= r^2 \sin^2 A (\cos^2 B + \sin^2 B)$$

$$x^2 + y^2 = r^2 \sin^2 A$$

Then  $x^2 + y^2 + Z^2 = r^2 \sin^2 A + Z^2$ 

$$r^2 = r^2 \sin^2 A + Z^2$$

$$Z^2 = r^2 (1 - \sin^2 A)$$

$$Z^2 = r^2 \cos^2 A$$

$$\Rightarrow$$
 Z = r cos A

**106.** Find the solutions for 'x' in eq.  $\frac{1}{a+b+x} = \frac{1}{a} + \frac{1}{b} + \frac{1}{x}$  is :

$$(1)$$
 -a, b

$$(2) -a, -b$$

$$(3) a, -b$$

$$(4)$$
 a, b

Ans. (2)

**Sol.** 
$$\frac{1}{a+b+x} = \frac{1}{a} + \frac{1}{b} + \frac{1}{x}$$

$$\Rightarrow \frac{1}{a+b+x} - \frac{1}{x} = \frac{1}{a} + \frac{1}{b}$$

$$\Rightarrow \frac{1}{a+b+x} - \frac{1}{x} = \frac{b+a}{ab}$$

$$\frac{x-a-b-x}{(a+b+x)x} = \frac{b+a}{ab}$$

$$\frac{-1}{(a+b+x)x} = \frac{1}{ab}$$

$$\Rightarrow$$
  $x^2 + (a + b)x + ab = 0$ 

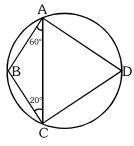
$$\Rightarrow$$
  $x^2 + ax + bx + ab = 0$ 

$$\Rightarrow$$
 x(x + a) + b (x + b) = 0

$$\Rightarrow$$
 (x + a) (x + b) = 0

$$\Rightarrow$$
 x = -a, -b

**107.** In the given figure, ABCD is a cyclic quadrilatical. If  $\angle BAC = 60^{\circ}$ ,  $\angle BCA = 20^{\circ}$  then find the value of  $\angle ADC$ ?



$$(2) 50^{\circ}$$

$$(3) 80^{\circ}$$

 $(4) 40^{\circ}$ 

Ans. (3)

**Sol.** In ∆ABC

$$60^{\circ} + \angle ABC + 20^{\circ} = 180^{\circ}$$
  
 $\Rightarrow \angle ABC = 100^{\circ}$ 

Since □ABCD is cyclic quadrilateral

So 
$$\angle ABC + \angle ADC = 180^{\circ}$$

$$\Rightarrow \angle ACD = 180^{\circ} - 100^{\circ}$$

$$\Rightarrow \angle ADC = 80^{\circ}$$

108.	A copper wire when bent in	the form of a square, enclo	oses an area of 484 cm². If the	e same wire is bent in the
	form of circle, the area enclo	sed by it is :		
	(1) 210 cm <sup>2</sup>	(2) 616 cm <sup>2</sup>	(3) 512 cm <sup>2</sup>	(4) 54 cm <sup>2</sup>

Ans. (2)

Then 
$$a^2 = 484$$

$$a = 22 \text{ cm}$$

According to question

$$4a = 2\pi r$$

$$2 \times 22 = \frac{22}{7} \times r$$
  $\Rightarrow r = 14 \text{ cm}$ 

area of circle = 
$$\pi r^2 = \frac{22}{7} \times 14 \times 14 = 616 \text{ cm}^2$$

**109.** The mean temperature of Monday to Wednesday was  $37^{\circ}$ C and of Tuesday to Thursday was  $34^{\circ}$ C. If the temperature on Thursday was 4/5th that of Monday. Then the temperature of Thursday was :

Ans. (4)

**Sol.** Let temperature from Monday to thursday are x, y, z, a respectively

Then 
$$\frac{x+y+z}{3} = 37$$
  
 $\Rightarrow x + y + z = 111$  ..... (1)  
And  $y + z + a = 3 \times 34$ 

$$y + z + a = 102$$
 .....(2)

Subtracting (2) from (1) then

$$x - a = 9$$

And 
$$a = \frac{4}{5}x \implies \frac{5a}{4} = x$$

So 
$$\frac{5a}{4} - a = 9$$

$$\Rightarrow \frac{a}{4} = 9$$

$$\Rightarrow$$
 a = 36°C

So temperature on thursady =  $36^{\circ}$ C

**110.** In a box containing 100 bulbs, 10 are defective. What is the probability that out of a sample of 5 bulbs none is defective.

$$(2)\left(\frac{1}{2}\right)^5$$

(3) 
$$\left(\frac{9}{10}\right)^{-5}$$

$$(4) \left(\frac{9}{10}\right)^5$$

Ans. (4)

**Sol.** Probability of getting nondefective bulbs =  $\frac{9}{10}$ 

In a sample of 5 bulbs probability of getting non defective =  $\left(\frac{9}{10}\right)^5$ 

111.	In IMF and World Bank the	decisions are taken by		
	(1) developing nation			
	(2) a joint council of develo	ped and developing coun	tries	
	(3) Western industrial power	ers		
	(4) least developed nations			
Ans.	(3)			
Sol.	In IMF and World Bank, the	e decisions are taken by W	estern Industrial Powers	
112.	What led to town planning	of the city of Bombay in $1$	880?	
	(1) The social tension		(2) Poverty	
	(3) Fear of plague epidemi	cs	(4) Riots	
Ans.	(3)			
Sol.	Fear of plague epidemics le	ed to the town planning of	the city of Bombay in 1880	
113.	Which of the following nov	el deals with caste oppress	ion?	
	(1) Sultana's Dream	(2) Indulekha	(3) Saraswati Vijayam	(4) Padmarag
Ans.	(3)			
Sol.	'Saraswati Vijayam' deals w	ith caste oppression		
114.	Who described Mazzini as '	the most dangerous enem	ny of our social order?	
	(1) Victor Emmanuel II	(2) Duke Metterinich	(3) Johan Gottfried	(4) Lord Byron
Ans.	(2)			
Sol.	Duke Metterinich described	l Mazzini as the most dang	gerous enemy of our social or	der
115.	Why was the Vietnam war	called the first television w	ar?	
	(1) Brought home, stories	from soldiers	(2) Led to increased sale	of teleivision sets
	(3) Battle scenes were show	wn on daily news	(4) Television was invente	d
Ans.	(3)			
Sol.	Vietnam war was called the	first television war as for	the first time Battle scenes w	ere shown on daily news
116.	What said "printing is the u	ltimate gift of God and the	e greatest one ?	
	(1) Charles Dickens		(2) J.V. Schely	
	(3) Mahatma Gandhi		(4) Martin Luther	
Ans.	(4)			
Sol.	Martin Luther said, "Printing	ng is the ultimate gift of Go	od and the greatest one"	
117.	Which of the following com	bination correctly indicate	s the three flows of internatio	onal economic exchange?
	(1) Capital, goods, raw ma	terial	(2) Goods, metal, labour	
	(3) Goods, labour, capital		(4) Labour, capital, food	grains
Ans.	(3)			
Sol.	Three flows of internationa	l economic exchange are	- Goods, Labour & Capital	
118.	The slogan 'Jai Hind' was g	iven by		
	(1) Lal Bahadur Shastri		(2) Subhash Chandra Bos	se
	(3) Jawahar Lal Nehru		(4) Ras Behari Bose	
Ans.	(2)			
Sol.	Slogan 'Jai Hind' was given	by Subhash Chandra Bose	e	

119.	what was the relationship o	i Guru Amardas ji with Gu	ru Kamaas Ji ?	
	(1) Father	(2) Son	(3) Brother	(4) Father-in-Law
Ans.	(4)			
Sol.	Guru Amardas ji was the Fat	ther-in-Law of Guru Ramd	as Ji	
<b>120</b> .	Name the treaty signed on 2	26th December, 1846 afte	r First Anglo Sikh War ?	
	(1) Treaty of Paris	(2) Treaty of Lahore	(3) Treaty of Bhayrowal	(4) Tripartite Treaty
Ans.	(2)			
Sol.	Treaty of Lahore was signed	l on 26th December, 1846	after First Anglo Sikh War	
121.	Who established Dal Khalsa	?		
	(1) Nawab Kapoor Singh		(2) Banda Bahadur	
	(3) Ranjit Singh		(4) Guru Gobind Singh Ji	
Ans.	(1)			
Sol.	Nawab Kapoor Singh establ	ished Dal Khalsa		
122.	'Relief', 'Cyclonic' and 'Conve	ectional' are types of which	of the following?	
	(1) Soil	(2) Water	(3) Rainfall	(4) Forests
Ans.	(3)			
Sol.	Relief, Cyclonic and Conven	itional are types of Rainfall		
<b>123</b> .	To which of the following, th	ne study of 'Seismology' is re	elated?	
	(1) Atmosphere	(2) Floods	(3) Tides	(4) Earthquakes
Ans.	(4)			
Sol.	Siesmology is related with E	arthquakes		
124.	What is the area of Chandig	arh?		
	(1) 114 sq kilometre	(2) 1014 sq kilometre	(3) 10140 sq kilometre	(4) 50362 sq kilometre
Ans.	(1)			
Sol.	114 sq.km is the area of Ch	andigarh		
125.	Which latitude passes appro	ximately through the midd	le of India ?	
	(1) Equator	(2) Tropic of Cancer	(3) Tropic of Capricon	(4) Prime Meridian
Ans.	(2)			
Sol.	Tropic of Cancer passes app	proximately through the mi	iddle of India	
<b>126</b> .	With which agent the featur	es like 'Stalactite' and 'Sta	lagmite' are related?	
	(1) Air	(2) Sea	(3) Underground water	(4) Glacier (River of Ice
Ans.	(3)			
Sol.	Stalactite and Stalagmite are	e the features developed by	y Underground water	
127.	In which state of India the 'K	andla Port' is situated?		
	(1) Punjab	(2) West Bengal	(3) Andhra Pradesh	(4) Gujarat
Ans.	(4)			
Sol.	Kandla Port is located in the	e state of Gujarat		
128.	Which of the following states	s has the least population d	ensity'?	
	(1) Bihar	(2) Uttar Pradesh	(3) Himachal Pradesh	(4) Arunachal Pradesh
Ans.	(4)			
Sol.	Arunachal Pradesh has the le	east population density		

129.	Which state of India has the	e maximum area under fore	ests?	
	(1) Haryana	(2) Tripura	(3) Rajasthan	(4) Karnataka
Ans.	(4)			
Sol.	Karnataka has the maximu	m area under forests		
130.	From which language the w	ordk 'Monsoon' has been d	lerived?	
	(1) Hindi	(2) Arabic	(3) German	(4) English
Ans.	(2)			
Sol.	Monsoon has been derived	from the Arabic language		
131.	For what mineral, the minim	ng region of 'Kolar' and 'Ra	mgiri in India are famous for	
	(1) Gold	(2) Silver	(3) Copper	(4) Bauxite
Ans.	(1)			
Sol.	Kolar and Ramgiri in India	are famous for Gold		
132.	Which state of India is conn	ected to China (Tibet Regio	on) through Nathula pass?	
	(1) Meghalaya	(2) Assam	(3) Sikkim	(4) Arunachal Pradesh
Ans.	(3)			
Sol.	Sikkim of India is connected	d to China through Nathula	Pass	
133.	Who was the Chairman of I	Orafting Commitee of India	n Constitution?	
	(1) Moti Lal Nehru	(2) Jawahar Lal Nehru	(3) B.R. Ambedkar	(4) Rajender Prasad
Ans.	(3)			
Sol.	B.R.Ambedkar was the Cha	airman of Drafting Commit	tee of Indian Constitution	
134.	Which of the following is a	big challenge to democracy	<i>;</i> ?	
	(1) Leaders	(2) Political partites	(3) Elections	(4) Illiterate Citizens
Ans.	` ,			
Sol.	Illiteracy is a big challenge	to democracy		
135.	Which of the following is no	ot the quality of ideal citizen	s?	
	(1) Good health	(2) Patriotism	(3) Tolerance	(4) Illiteracy
Ans.	(4)			
	Illiteracy is the not the quali	•		
136.	Seats are reserved for wom	nen in		
	(1) Parliament		(2) Panchayati Raj Institution	1
	(3) State Legislature		(4) Rajya Sabha	
Ans.	` '			
	Seats are reserved for wom	•	utions	
137.	Which type of government	_		
	(1) Centre and State Gover		(2) Govt. of two States	
	(3) Centre and two State C	Sovernment	(4) Centre and three State (	Government
Ans.	• •			
	Centre and State Governm	_	ts.	
138.	How many state government			
	(1) 7	(2) 28	(3) 13	(4) 29
Ans.	• •			
Sol.	29 state governments are for	unctioning in India		

139.	An ordinary bill is represent	ed in		
	(1) Rajya Sabha		(2) Lok Sabha	
	(3) Any house of Parliament	:	(4) Rajya Vidhan Sabha	
Ans.	(3)			
Sol.	An ordinary bill can be intro	oduced in any house of Parli	ament	
140.	Who is the constitutional he	ad of India		
	(1) King		(2) Queen	
	(3) Prime Minister		(4) President	
Ans.	(4)			
Sol.	President is the constitution	al head of India		
141.	Which of the following is no	t the National Flag ?		
	(1) Shriomany Akali Dal		(2) Bhartiya Janta Party	
	(3) Indian National Congres	SS .	(4) Bahujan Samaj Party	
Ans.	(1)			
Sol.	Shiromany Akali Dal is a re	gional party, not a national	political party	
142.	India is a federal state beca	use		
	(1) Dual judiciary		(2) Written constitution	
	(3) Dual citizenship		(4) Share of power between	centre and state
Ans.	(4)			
Sol.	India is a federal state becar	use it shares power betweer	n Central and State governme	nt
143.	GDP can be written as			
	(1) General Domestic Prod	uct	(2) Gross Domestic Product	
	(3) Gross Development Pro	oduct	(4) General Development Pr	oduct
Ans.	(2)			
Sol.	GDP stands for Gross Dom	estic Product		
144.	Foreign Trade			
	(1) Increases choice of good	ds	(2) Decreases price of goods	
	(3) Increases competition in	n the market	(4) Decreases earnings	
Ans.	(3)			
Sol.	Foreign trade increases con	npetition in the market		
145.	COPRA full form is			
	(1) Co-ordination Protection	n Act	(2) Co-education Protection A	Act
	(3) Co-operative Protection	Act	(4) Consumer Protection Act	
Ans.	(4)			
Sol.	Full form of COPRA is Cons	sumer Protection Act		
146.	Which of these notes are iss	sued by finance department	of Govt. of India	
	(1) ₹ 500	(2) ₹ 100	(3) ₹ 2000	(4) ₹ 1
Ans.	(4)			
Sol.	One Rupee Note is issued b	y finance department of Go	vernment of India	

147.	What is the impact of green	revolution		
	(1) Forest area increased		(2) Air Pollution decreased	
	(3) Wheat Production Increa	ased	(4) Milk Production Increased	d
Ans.	(3)			
Sol.	Wheat production increased	l as a impact of Green Revo	olution	
148.	Globalisation was stimulated	by		
	(1) Money	(2) Transportation	(3) Population	(4) Computers
Ans.	(2)			
Sol.	Globalisation was stimulated	by Transportation		
140	Λ			
149.	A consumer		<del></del>	
149.	(1) Sells goods and services		(2) Buy goods and services	
149.				
Ans.	<ul><li>(1) Sells goods and services</li><li>(3) Produces goods and services</li></ul>		(2) Buy goods and services	
Ans.	<ul><li>(1) Sells goods and services</li><li>(3) Produces goods and services</li></ul>	vices.	<ul><li>(2) Buy goods and services</li><li>(4) Computers</li></ul>	
Ans. Sol.	<ul><li>(1) Sells goods and services</li><li>(3) Produces goods and services</li><li>(2)</li></ul>	vices. services is termed as Const	<ul><li>(2) Buy goods and services</li><li>(4) Computers</li></ul>	
Ans. Sol.	<ul><li>(1) Sells goods and services</li><li>(3) Produces goods and services</li><li>(2)</li><li>Person who buys goods and</li></ul>	vices. services is termed as Const	<ul><li>(2) Buy goods and services</li><li>(4) Computers</li></ul>	(4) 2017-22
Ans. Sol.	<ul> <li>(1) Sells goods and services</li> <li>(3) Produces goods and services</li> <li>(2)</li> <li>Person who buys goods and</li> <li>What is the time-period of 1</li> <li>(1) 2002-07</li> </ul>	vices. services is termed as Const 2th five year plan?	(2) Buy goods and services (4) Computers  umer	(4) 2017-22